

Melanoides tuberculata as Bioindicator of the Heavy Metal Contamination in Water and Sediment Pollution of Euphrates River at Thi-Qar province, Iraq

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ABSTRACT

Aims The aim of this study was determination the diversity of the aquatic Gastropoda species and to assess heavy metals in water, sediment, soft tissue, and shell *Melanoides tuberculata* by evaluating some of the physical and chemical factors for the river.

Material & Methods The study was done during the Summer and Winter seasons and samples were collected from three study sites along the Euphrates river. The samples were analyzed using Atomic Absorption Spectroscopy for the detection of heavy metals.

Findings The results revealed that *Melanoides tuberculata*, *Physella acuta*, *Pisidium dubium*, *Melanopsis nodosa*, and *Corbicula fluminea* were found in the studied station in both seasons with a high percent ratio (37.73, 20.27, 6.58, 8.42, 26.98%, respectively) in the summer season. Sediment has the highest mean concentration more than water. The heavy metals order was arranged as Zn<Ni<Pb<Cd and Zn<Ni<Cd<Pb for water and sediment respectively. Station 2 has the highest concentration of metals during the study. The concentration of metals in soft tissue in *Melanoides tuberculata* is higher than in shell and they were arranged as Pb<Zn<Ni<Cd for soft tissue and Ni<Zn<Pb<Cd for the shell.

Conclusion Lead and Zink were higher and exceed the permissible level. *Melanoides tuberculata* could accumulate the heavy metals in their metals then it could use as a bio-indicator for these metals to detect the pollution in the Euphrates River.

Keywords Gastropoda; Heavy Metals; Bioindicator

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Introduction

Water is one of the renewable sources that represent the base stone of the sustainability of human life and all kinds of living creatures on the surface of the earth [1]. The total water on the Earth's surface represents about 71%. Available freshwater represents 3%, of which available for human use constitutes only 0.4% [2]. Water is home to many trace minerals found in the oceans, where more than 70 elements have been confirmed [3]. The interest in studying the physical and chemical factors of aquatic environments, including rivers, is due to the possibility of these factors in these ecosystems create suitable conditions for the biological diversity of living organisms and the extent of changes that occurred due to human activity them from releases of organic carbon, nutrients and harmful substances, and their content of dissolved oxygen (DO) in them [4].

Heavy metals from anthropogenic pollution sources are always ejected into aquatic and terrestrial ecosystems therefore, the concern was increased to evaluate the effect of man pollution on a different kind of ecosystem is growing. Pollution with heavy metals is a hazard because of their poisonousness, bioaccumulation, and bio magnifications in the food chain [5]. Increasing heavy metals concentration can cause harmful and direct physiological toxic damage when they are deposited in tissues. Heavy metals are categorized into essential and non-essential groups and these groups convert to a toxic state at high intakes level [6]. The dispersal of industrialization, increase of population, and using serval technical devices of agriculture, have presented various pollutants into the aquatic ecosystems [7]. Heavy metals are released from manufacturing and transferred into the environment then concentrate and bio-amplified along the food chain [8].

The aquatic environment has a critical role in the dispersal and installation of living organisms communities signified by fish, macroinvertebrates, and other aquatic organisms [9].

The biodiversity of freshwater mollusks amounts to nearly 5,000 species, and Freshwater gastropods include about 4, 000 species [10], while the number of bivalves is approximately 1200 species [11].

Mollusks are a food source for fish, birds, and other invertebrates, and humans use them as a food source in many countries. Mollusks have a role in cycling the elements in nature and purifying water, so the presence of mollusks is a good vital indicator of environmental quality in all aquatic habitats [12]. Although mollusks are a food source for humans, they may pose a danger to them through the ability of these invertebrates to accumulate toxins inside their bodies in certain circumstances and thus expose humans to poisoning as a result of their consumption [13].

Melanoides tuberculata is famous as one of two universally invasive gastropods and it spreads through East Africa, the Middle East, and Southeast Asia. *Melanoides tuberculata* naturally lives in a wide diversity of freshwater or brackish water including rivers, streams, and wetlands [14, 15]. it is a cosmopolitan species from the class Gastropoda which has light brown color and elongated " shells higher than wide, and it is a cosmopolitan species [16]. The snail has an operculum that can guard snail from dryness. This snail feeds mainly on microalgae and performances as an intermediate host for many trematodes [17].

It is a considerable food source for fish, some small carnivores, and water birds [18, 19], Moreover, they are intermediate predictors of many species of parasites that infect humans and animals. Furthermore, *Melanoides tuberculata* has features like a high growth rate, the capacity for migration, and territorial expansion. *M. tuberculata* is the greatest wide-ranging species of the family Thiariidae, which can live in practically any sort of freshwater causing this species one of the most persistent macroinvertebrates [20].

The current study aimed to determine and evaluated the heavy metals in water, sediment, and *Melanoides tuberculata* and their relationship with some physical and chemical factors.

Material and Methods

In this experimental research, the trace metal content (Cadmium: Cd, Lead: Pb, Nickle: Ni, Zink: Zn) was investigated in water, sediment, and the soft tissue and shell of *Melanoides tuberculata*. The Euphrates river is the lengthiest in Western Asia, the river initiates from Turkey and transfers to Syria, then arrives in Iraq and connects with the Tigris river in Basra city southern Iraq. The river is about 2786 km [21]. Four stations in the Euphrates river in Nasiriyah city were studied.

- 1- Station 1 is located at the beginning of the river when entering Nasiriyah city.
- 2- Station 2 is located near the thermal electric power station in the junction zone of hot water emerging with the river.
- 3- Station 3 is located far away from the second station when the river leaves the city of Al-Nasiriyah.

Measurements were made for all study sites for only two different seasons in terms of climatic conditions, the winter and summer seasons. The samples were collected using clean polyethylene bottles (Glaswarenfabrik Karl Hecht; Germani). Sampling was analyzed for some physical properties and checked immediately after collection. The air and water temperature were measured using a graduated mercury thermometer U14295 (3B Scientific GmbH; USA) from 0-100°C, while the pH

concentration and electrical conductivity (EC; $\mu\text{S}/\text{cm}^1$) were measured using a Multimeter device (EC-PCSTestr35; Hana Inc). In addition, DO was measured using the method described in a study [22]. The mollusk samples were also collected using a manual sieve with a diameter of 0.5mm, where the collected samples are washed with water to get rid of the impurities stuck in them, after which they are diagnosed and classified according to studies [23, 24]. Sediment samples, an exact weight of the dry sediment sample (0.5g) was fully digested in Teflon vessels using a mixture of concentrated acids (HNO_3 and HClO_4 (AnalaR; Australia) for about 2 hours (triplicate digestion was made for each sample). The final solution was diluted to 25 ml with distilled de-ionized water. Their metal content was measured by AAS described by [25].

Sediments were analyzed by collecting samples using cleaned and acid-washed PVC corer and immediately kept in pre-cleaned and acid-washed polythene bags, which were sealed and kept in an ice box until further analysis in the laboratory. Sediment samples were washed with metal-free double distilled water. The sediment samples were then dried in an oven at 60°C for 5-6 hours. Dried sediment samples were ground in a glass mortar and reduced into fine particles. Metals were analyzed through atomic absorption spectrophotometer; 2.0g from soft tissues were weighed for mollusk then 1ml of digestion solution (concentrated nitric acid, concentrated sulfuric acid, and concentrated perchloric acid, with a percentage of 1:1:1) was added to flasks of content. The flasks closed strongly and left for 48-72 hours at room temperature to finish the digestion process. Then, completed to 10ml with deionized water and filtered. Finally, measured Atomic Spectrophotometric Absorption (Byahut Scientico; Iniamart; India).

Findings

Some physical and chemical parameters and heavy metals concentrations for water were studied in Tables 1 and 2. The result of the present study showed that the maximum mean concentration was reordered in summer and the minimum was in winter, the temperature of the air was higher than water, and station 2 recorded a high value for both air and water temperature. For pH, all the mean showed alkaline nature, and the average was arranged between 7.80-8.40 and slight fluctuations in the values of pH appeared in along the study period. The EC mean in the aquatic location ranged from $1945.00\text{--}3319.00\mu\text{S}/\text{cm}^1$. The highest mean was recorded in winter at station 2 whereas the lowest mean was recorded in summer at station 3. For DO mean concentration was arranged from 5.00-7.12. Winter has recorded the highest mean at

stations 2 and 3 while the lowest mean was recorded in summer at station 1.

The mean concentration for some heavy metals in water was recorded (Table 2). The lowest and highest mean concentration was arranged between 0.06-0.11, 0.16-0.25, 3.17-6.16, 33.88-56.14 $\mu\text{g}/\text{l}$ for Cd, Pb, Ni, and Zn metals respectively. The result showed that station2 recorded the highest value of the mean concentration of metals. The descending order of heavy metals was: $\text{Zn}<\text{Ni}<\text{Pb}<\text{Cd}$. The was variation between seasons, summer has the highest mean concentration of heavy metals more than winter.

Table 1) Mean results of some physical and chemical parameters for water in studied stations of Euphrates river at Thi-Qar

Station	Air Temp. ($^\circ\text{C}$)	Water Temp. ($^\circ\text{C}$)	pH	EC ($\mu\text{S}/\text{cm}^1$)	DO (mg/L)
Winter					
1	12.20 \pm 0.10	9.00 \pm 0.00	8.22 \pm 0.00	3234.00 \pm 2.00	6.99 \pm 0.04
2	12.00 \pm 0.00	12.00 \pm 0.10	8.22 \pm 0.01	3319.00 \pm 4.00	7.12 \pm 0.00
3	12.90 \pm 0.10	9.10 \pm 0.10	8.40 \pm 0.01	3109.00 \pm 1.00	7.12 \pm 0.01
Total	12.36 \pm 0.41	10.03 \pm 1.47	8.28 \pm 0.09	3220.66 \pm 91.50	7.07 \pm 0.06
Summer					
1	49.30 \pm 0.10	41.00 \pm 0.10	7.90 \pm 0.01	2391.00 \pm 6.00	5.00 \pm 0.00
2	49.00 \pm 0.00	42.00 \pm 0.10	8.00 \pm 0.00	2073.00 \pm 2.00	6.00 \pm 0.00
3	48.80 \pm 0.10	41.10 \pm 0.10	7.80 \pm 0.10	1945.00 \pm 1.00	6.40 \pm 0.01
Total	49.03 \pm 0.22	41.36 \pm 0.48	7.90 \pm 0.10	2136.33 \pm 198.90	5.80 \pm 0.62
LSD	0.20 \pm 0.00	0.20 \pm 0.00	0.20 \pm 0.00	0.08 \pm 0.00	0.04 \pm 0.00

Table 2) Mean concentration of heavy metals ($\mu\text{g}/\text{l}$) for water in Euphrates river at Thi-Qar proven

Station	Cd	Pb	Ni	Zn
Summer				
1	0.06 \pm 0.01	0.17 \pm 0.00	3.17 \pm 0.02	33.88 \pm 0.29
2	0.11 \pm 0.01	0.25 \pm 0.01	6.16 \pm 0.06	56.14 \pm 0.83
3	0.06 \pm 0.00	0.16 \pm 0.01	4.22 \pm 0.00	39.72 \pm 0.48
Total	0.07 \pm 0.02	0.19 \pm 0.04	4.51 \pm 1.31	43.25 \pm 10.00
LSD	0.04 \pm 0.00	0.05 \pm 0.00	0.15 \pm 0.00	2.33 \pm 0.00
Winter				
1	0.07 \pm 0.00	0.19 \pm 0.00	4.12 \pm 0.01	35.02 \pm 0.26
2	0.08 \pm 0.00	0.25 \pm 0.00	6.13 \pm 0.02	53.19 \pm 0.95
3	0.07 \pm 0.00	0.16 \pm 0.00	4.32 \pm 0.00	39.67 \pm 0.57
Total	0.12 \pm 0.02	0.20 \pm 0.04	4.86 \pm 0.95	42.63 \pm 8.19
LSD	0.01 \pm 0.00	0.01 \pm 0.00	0.06 \pm 0.00	2.64 \pm 0.00

The results of Table 3 showed the mean concentration and the stander deviation for heavy metals in sediments. The mean value in sediments ranged between 0.11-0.14, 8.16-10.49, 45.73-65.76, and 37.87-59.19 $\mu\text{g}/\text{g}$ for Cd, Pb, Ni, and Zn metals, respectively. The metals were arranged in order: $\text{Zn}<\text{Ni}<\text{Cu}<\text{Pb}$. The highest mean concentration was at station 2 in summer while the lowest mean was in winter. The mean concentration in summer was higher than in the winter.

The highest frequency of the different species of mollusks in studied stations was recoded for *Melanoides tuberculata*, *Corbicula fluminea*, and *Physella acuta*, respectively; While the lowest frequency was for *Melanopsis nodosa* and *Pisidium dubium* in summer and winter, respectively. *Melanoides tuberculata* more dispersal and had the highest value in summer in the studied area (Table 4).

Table 3) Mean concentration of heavy metals ($\mu\text{g/g}$) for sediment in Euphrates river at Thi-Qar proven

Station	Cd	Pb	Ni	Zn
Summer				
1	0.12 \pm 0.00	8.16 \pm 0.02	45.73 \pm 0.52	37.87 \pm 0.00
2	0.14 \pm 0.00	9.91 \pm 0.02	60.61 \pm 0.33	44.16 \pm 0.00
3	0.11 \pm 0.00	9.22 \pm 0.05	47.22 \pm 0.00	41.41 \pm 0.58
Total	0.12 \pm 0.01	9.09 \pm 0.76	51.19 \pm 7.10	41.14 \pm 2.74
LSD	0.02 \pm 0.00	0.1 \pm 0.00	1.43 \pm 0.00	1.34 \pm 0.00
Winter				
1	0.12 \pm 0.00	9.34 \pm 0.00	51.12 \pm 0.00	45.00 \pm 0.00
2	0.14 \pm 0.00	10.49 \pm 0.02	65.76 \pm 0.52	59.19 \pm 0.89
3	0.12 \pm 0.00	10.01 \pm 0.00	55.00 \pm 1.14	50.41 \pm 0.42
Total	0.12 \pm 0.00	9.94 \pm 0.49	57.29 \pm 6.59	51.53 \pm 6.22
LSD	0.01 \pm 0.00	0.06 \pm 0.00	2089 \pm 0.00	2.27 \pm 0.00

Table 4) Statistic estimation of the aquatic Gastropod from three stations in Euphrates river at Thi-Qar proven

Identified Species	Number	Percentage	Statistic Estimation
Summer			
<i>Melanoides tuberculata</i> (Müller, 1774)	860	37.73	$\chi^2 = 771.97$; df= 4; p-value= 0.001
<i>Physella acuta</i>	462	20.27	
<i>Pisidium dubium</i>	150	6.58	
<i>Melanopsis nodosa</i> (Oliver, 1804)	192	8.42	
<i>Corbicula fluminea</i>	615	26.98	
Total	2279	100	
Winter			
<i>Melanoides tuberculata</i>	152	28.25	$\chi^2 = 64.361$; df= 4; p-value= 0.001
<i>Physella acuta</i>	129	23.97	
<i>Pisidium dubium</i>	51	9.47	
<i>Melanopsis nodosa</i>	78	14.98	
<i>Corbicula fluminea</i>	128	23.79	
Total	538	100	

The mean concentration and slandered deviation for heavy metals in soft tissue and shell of *Melanoides tuberculata* showed in tables 5 and 6. Results revealed that the mean concentration of heavy metals was higher in soft tissue than in shells during the summer season. The mean concentration of Pb was high about 0.40-014 $\mu\text{g/g}$ in station 2 during summer in soft tissue and shell respectively; while Zn was 29.14-5.22 in station 2 during summer. The metals were arranged in order: Pb<Zn<Ni<Cd for soft tissue and Ni<Zn<Pb<Cd for the shell.

Table 5) Mean concentration of heavy metals ($\mu\text{g/g}$) for soft tissue in *Melanoides tuberculata*

Station	Cd	Pb	Ni	Zn
Winter				
1	0.05 \pm 0.00	0.38 \pm 0.01	8.91 \pm 0.01	24.18 \pm 0.01
2	0.05 \pm 0.00	0.40 \pm 0.00	9.84 \pm 0.00	29.14 \pm 0.02
3	0.03 \pm 0.00	0.38 \pm 0.00	8.92 \pm 0.02	26.97 \pm 0.01
Total	0.04 \pm 0.01	0.39 \pm 0.01	9.22 \pm 0.46	26.76 \pm 2.15
LSD	0.02 \pm 0.00	0.03 \pm 0.00	0.09 \pm 0.00	0.06 \pm 0.00
Summer				
1	0.04 \pm 0.00	0.31 \pm 0.00	7.72 \pm 0.01	20.22 \pm 0.01
2	0.05 \pm 0.00	0.35 \pm 0.01	7.91 \pm 0.00	22.24 \pm 0.05
3	0.03 \pm 0.00	0.36 \pm 0.00	7.44 \pm 0.01	22.20 \pm 0.01
Total	0.04 \pm 0.01	0.34 \pm 0.02	7.69 \pm 0.02	21.55 \pm 0.99
LSD	0.01 \pm 0.00	0.03 \pm 0.00	0.04 \pm 0.00	0.13 \pm 0.00

Table 6) Mean concentration of heavy metals ($\mu\text{g/g}$) for shell in *Melanoides tuberculata*

Station	Cd	Pb	Ni	Zn
Winter				
1	0.010 \pm 0.00	0.08 \pm 0.01	13.86 \pm 0.00	2.69 \pm 0.24
2	0.023 \pm 0.00	0.12 \pm 0.01	14.67 \pm 0.05	3.77 \pm 0.20
3	0.011 \pm 0.00	0.11 \pm 0.01	13.11 \pm 0.00	3.27 \pm 0.29
Total	0.014 \pm 0.00	0.10 \pm 0.02	13.88 \pm 0.67	3.24 \pm 0.51
LSD	0.01 \pm 0.00	0.04 \pm 0.00	0.1 \pm 0.00	0.9 \pm 0.00
Summer				
1	0.033 \pm 0.00	0.11 \pm 0.01	15.56 \pm 0.57	4.82 \pm 0.38
2	0.045 \pm 0.00	0.14 \pm 0.02	19.44 \pm 0.57	5.22 \pm 0.03
3	0.006 \pm 0.00	0.14 \pm 0.00	17.78 \pm 0.62	4.23 \pm 0.22
Total	0.028 \pm 0.01	0.13 \pm 0.01	17.59 \pm 1.76	4.76 \pm 0.48
LSD	0.004 \pm 0.00	0.05 \pm 0.00	2.35 \pm 0.00	1.02 \pm 0.00

Discussion

The physical and chemical parameters of water are an essential factor for living the aquatic organism and affect their distribution of them [26]. Temperature is the most important factor for the living organism and influenced many processes like reproduction and distribution in both animals and plants [27]. The results showed that the temperature of the water was less than air, and water in both seasons cause they were affected by solar radiation from the sun and the water was shallow in that area [28]. That relationship made great conjunction between air and water temperature. For pH, the average was arranged between 7.80-8.40 which made the water slightly alkaline. Iraq water was tend to be alkaline [29]. The alkane was formed by the photosynthetic rate which depended on respiration and decomposition [30]. The electrical conductivity in the aquatic location is related to the soluble solids in the water therefore, it formed an excellent parameter to appear the purity of the water [31]. The value of electrical conductivity was 3220.66 in winter and 2136.33 in summer. The rising of the value in winter was related to the autumn rain during winter which increasing the washing of soil near the river.

The DO ratio was arranged from 5.00-7.12 and the DO value was high in winter than summer season because the increase in temperature decreases the DO. The high level of DO appears of good water quality [26]. Many factors affect the DO value like temperature and salinity and the number of abiotic [32].

Heavy metals are one of the more serious pollutants in our natural environment due to their toxicity, persistence, and bioaccumulation problems [33]. Heavy metals are pollutants that can yield huge harm to the atmosphere once they have high concentrations [19-34].

The present study showed clear differences in heavy metal concentrations between studied stations and seasons. The concentrations of heavy metals were higher in summer at station 3. This was due to the

exposure of station 3 to several forms of contamination such as chemicals, waste, oil spilled, and sewage that were thrown from the thermal electric power station. Besides the presence of residential areas near this station.

Moreover, there were higher concentrations of studied metals in the summer than in the winter. This happened because of water evaporation which results from increasing temperatures and then increasing concentrations of heavy metals. Furthermore, degradation of organic matter could increase the heavy metals concentration [35] these results were agreed with studies [36-38].

Sediment formed an essential part of the river, it made a sink for it and most heavy metals settled on it. The current study revealed that sediment has a high concentration of heavy metals than water table 3. The results of the study were shown that the concentration of studied metals in sediment was higher than their concentration in water except for Pb. This happened due to deposit of metals with the suspended matter in surface water at the bottom and increased with the ability of organic matter to adsorption [39, 40]. Station 2 has a high concentration of heavy metals compared to the other stations because of its location near the population area which throws the waste that increased the organic matter and then increases the absorption of heavy metals. Many factors affect the concentration of heavy metals in sediments such as pH, degradation of plankton, Diatoms, and organisms. Moreover, temperature, salinity, the proportion of organic matter in sediments, and sediment grain size were the most important factor that affects the concentration of heavy metals in sediment [41]. The result agreed with [42, 43]. The present study showed a high concentration of Pb at station 2 in the summer is related to the high traffic density near this area of Al-Nassiriya city.

The result in table 4 was showed the percentage of the different species of mollusks in the studied stations. The highest presence was recorded for *Melanoides tuberculata* (37.7-152%), *Corbicula fluminea* (26.98-128%) then *Physella acuta* (20.27-23.78%) while the lowest percentages were *Melanopsis nodosa* (8.2-1.98%) and *Pisidium dubium* (6.58-9.47%) for summer and winter respectively. The current study showed an increase in the appearance of some species during summer more than winter cause the increasing temperature could increase the metabolic activity of Mollusca. Furthermore, the physical and chemical parameters for water affect the growth and diversity of aquatic invertebrates [44]. *Melanoides tuberculata* has the highest ratio in the Euphrates river cause these species prefer the lotic water. All studied species were found during both seasons and this disagrees with research that showed that many species of Mollusca die in the winter or low-temperature degrees [45, 46]. Interestingly, the rising temperature

formed an essentially important part in the proliferation of snails. Our study is constant with the study [29].

The present study revealed that the concentration of heavy metals in snail tissue and the shell was higher in summer than in winter at station 2 Table [4, 5]. Interestingly the concentration of Pb and Ni in tissue was less than in the shell. It is well known that the uptake of trace metals by an aquatic organism is dependent on the metal concentration in the solution until situated [47]. Therefore, the heavy metals concentration in the snail was more than in water cause the organic matter keep the metals the snail absorb. The level of accumulation in the organism was changed according to the ambient concentration and environmental concentrations of metals [48].

The Pb concentration in the studied snail was high especially in sediment and soft tissue which exceeded the presumable level and made 0.30 mg/kg set by [49]. The high concentration can produce a possible risk of toxicity and Pb poisoning. Zn is a vital nutrient for all organisms. The concentrations of Zn were high in all studied samples but within the FAO [50] reference limit of 130.0 mg/kg. increasing of intake Zn from organisms having a high concentration of Zn resulting in vomiting, nausea, headaches, and diarrhea. Inflowing of wastewater from the thermal station and area near station 2 was the main source of raising the concentration of Zn. The result of the present study agreed with [51, 52]. Almost of the heavy metal concentrations exceed the permissible level of WHO [53].

Conclusion

This study declared that the concentration of heavy metals concentration is elevated in the specific station which has waste from human activities and these added levels of metals to the river. So attention should increase to monitoring the level of heavy metals and use organisms like snails and shrimp as bio-indicator.

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